

Section: Sat, March 20, 2021 10am-
10:50am

INTRODUCTION TO COMPUTER- AIDED DRUG DESIGN (CADD)

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IS THIS THE CLASS YOU ARE LOOKING FOR?

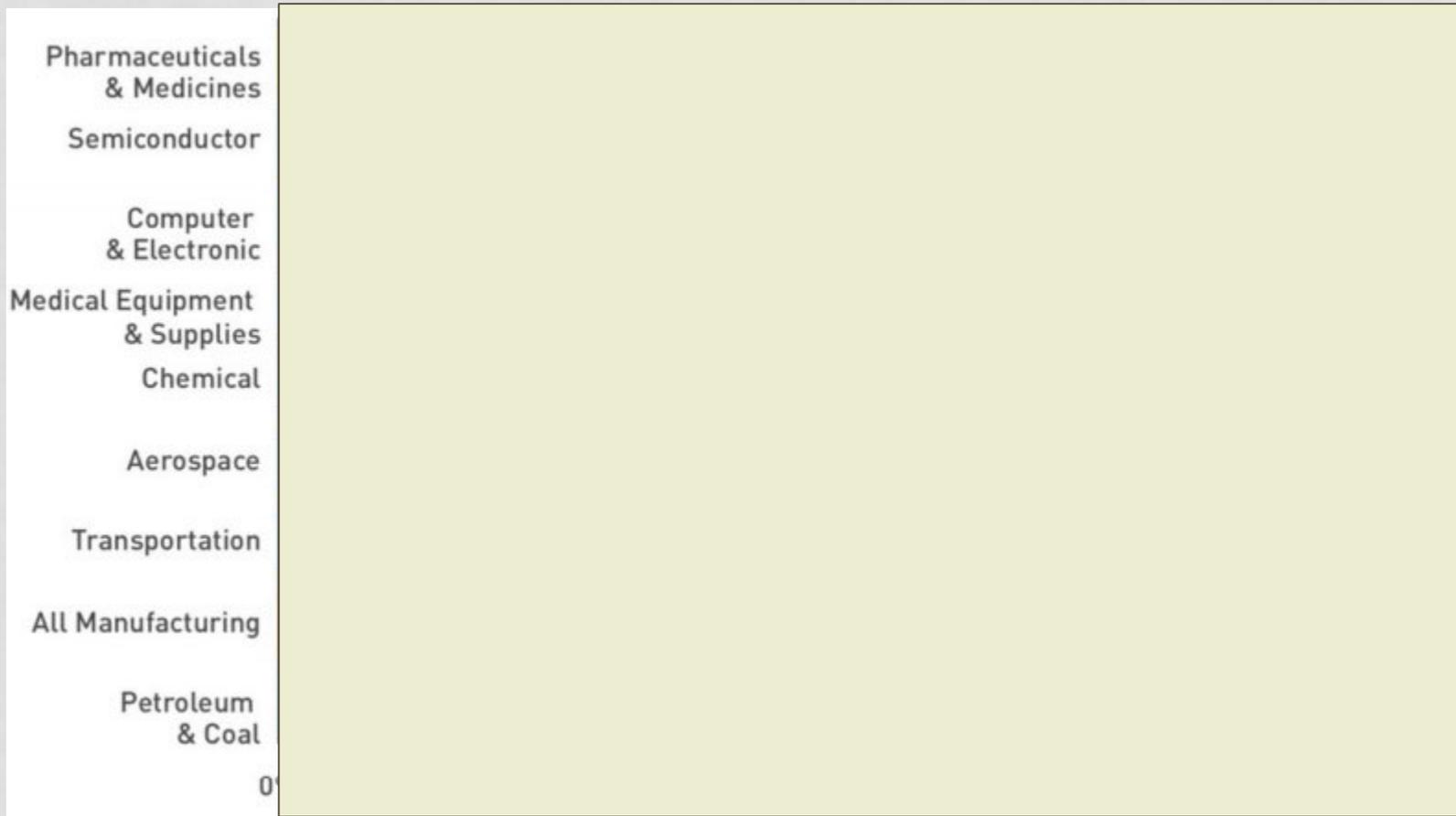
- Computer-aided drug design (CADD) uses the tools of computational chemistry and data science to aid the process of drug (medication; not the bad kind) discovery.
- Almost everything you learned about the lock and key model of protein-ligand binding is wrong. Concepts include,
 - the drug discovery pipeline and associated costs,
 - ligand-based vs. structure-based drug design,
 - lock and key vs. allostery
 - hit Identification and lead optimization,
 - Lipinski's rule of five,
 - structure-function relationships,
 - high-throughput virtual screening,
 - agonist vs. antagonist vs. inverse agonist,
 - decoys,
 - selectivity and side effects,
 - drug docking
 - molecular dynamics
 - more advanced computational methods.

- **Prerequisites**

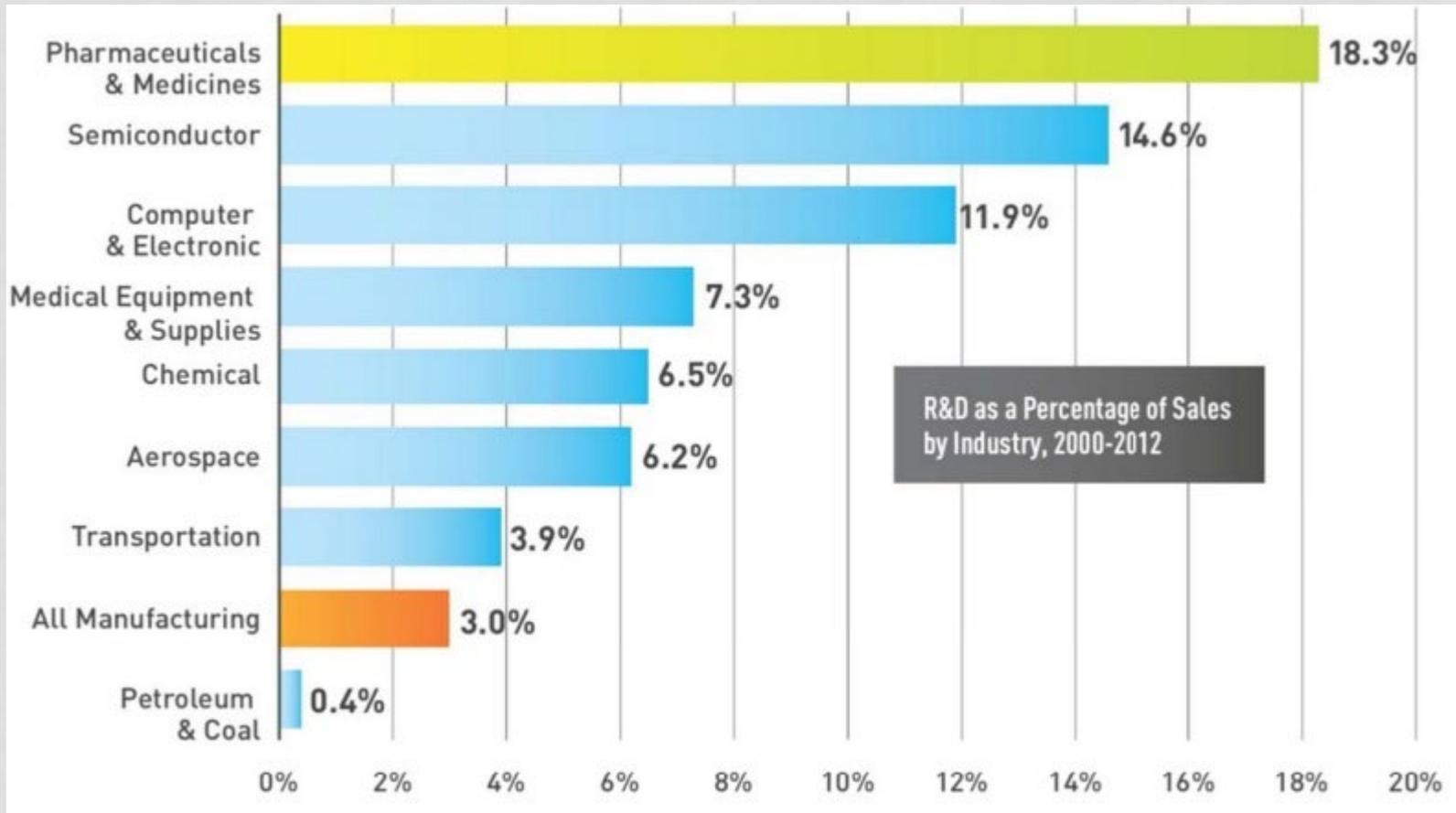
High school level biology/chemistry in grades 10-12

AP Biology and/or Chemistry are/is a huge plus.

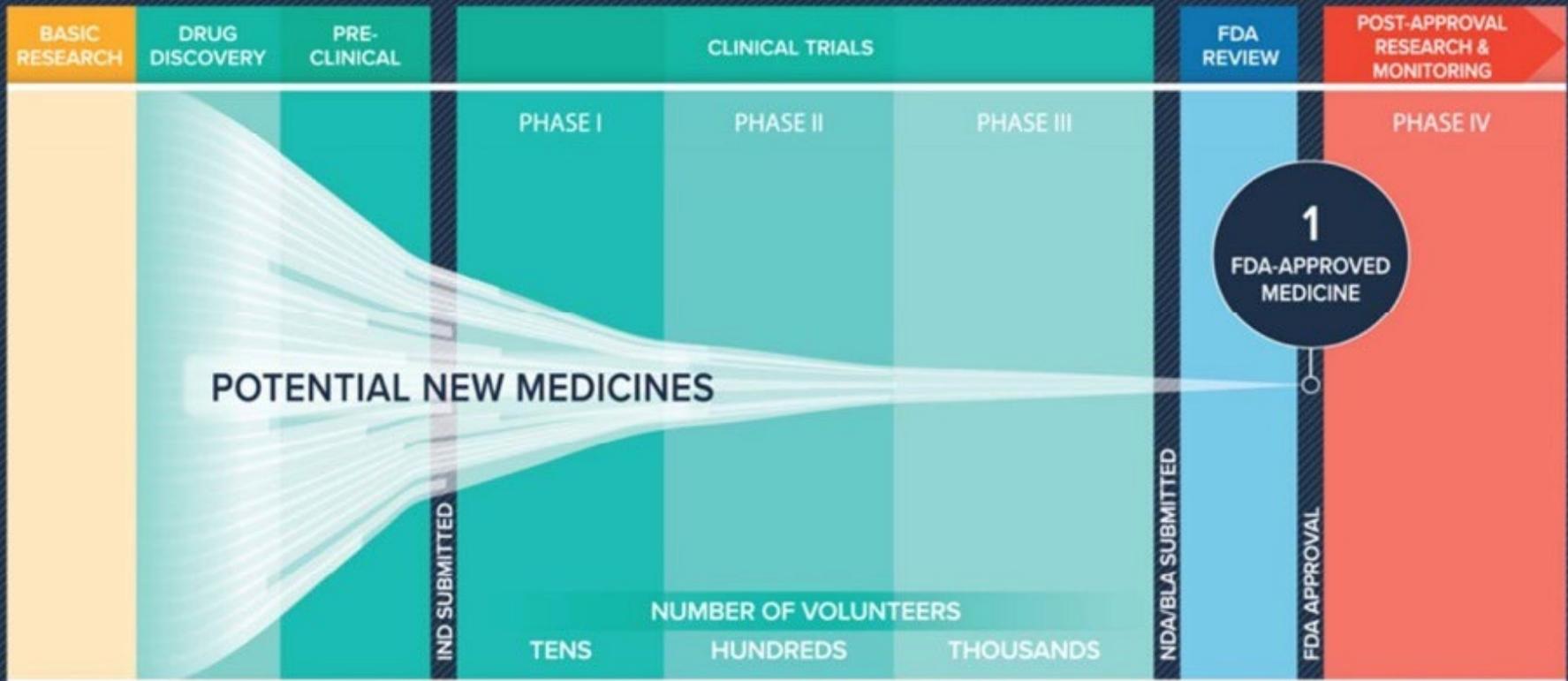
Costs of Pharmaceuticals



Costs of Pharmaceuticals

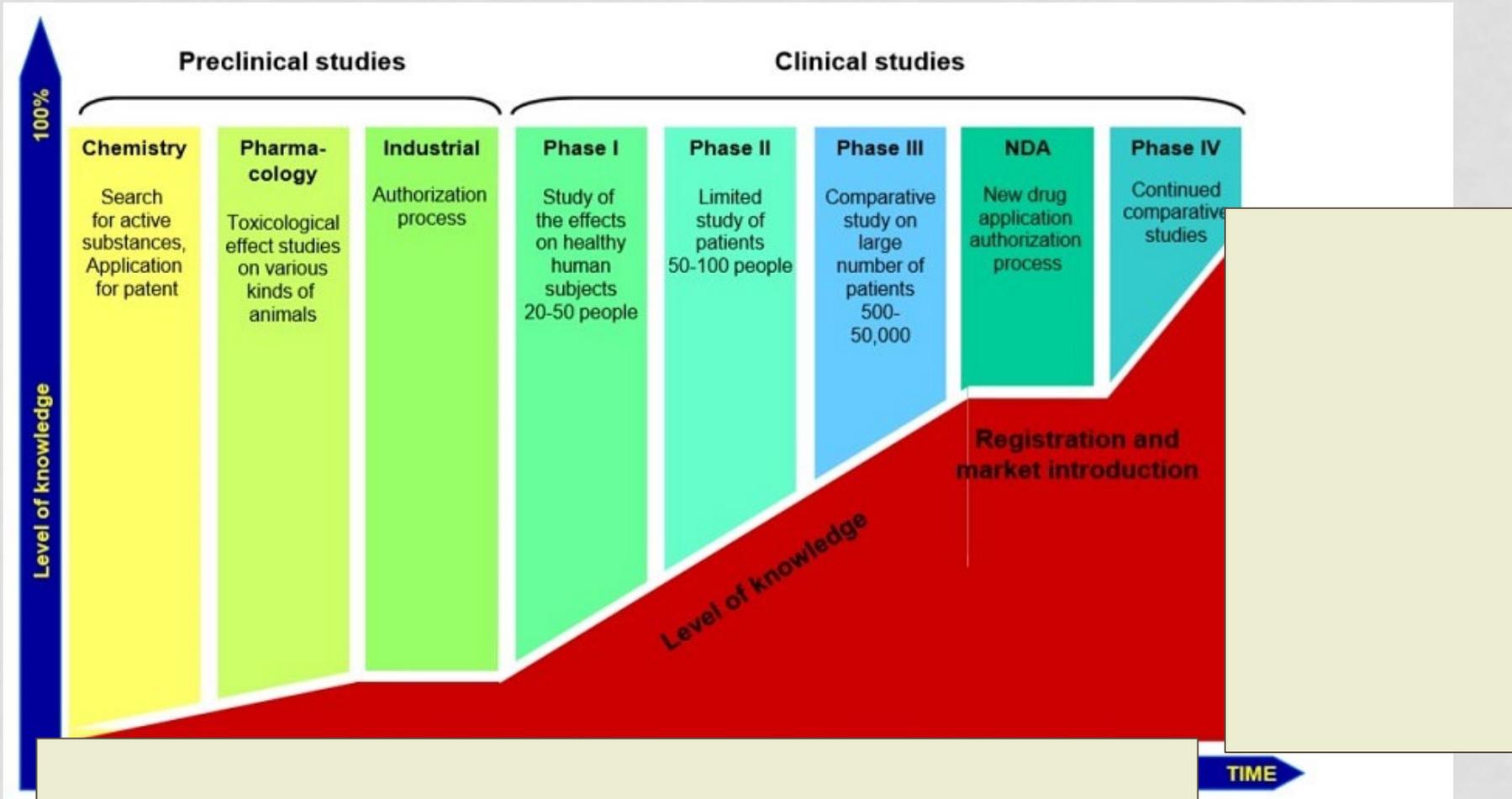


Fail Fast, Fail Cheap

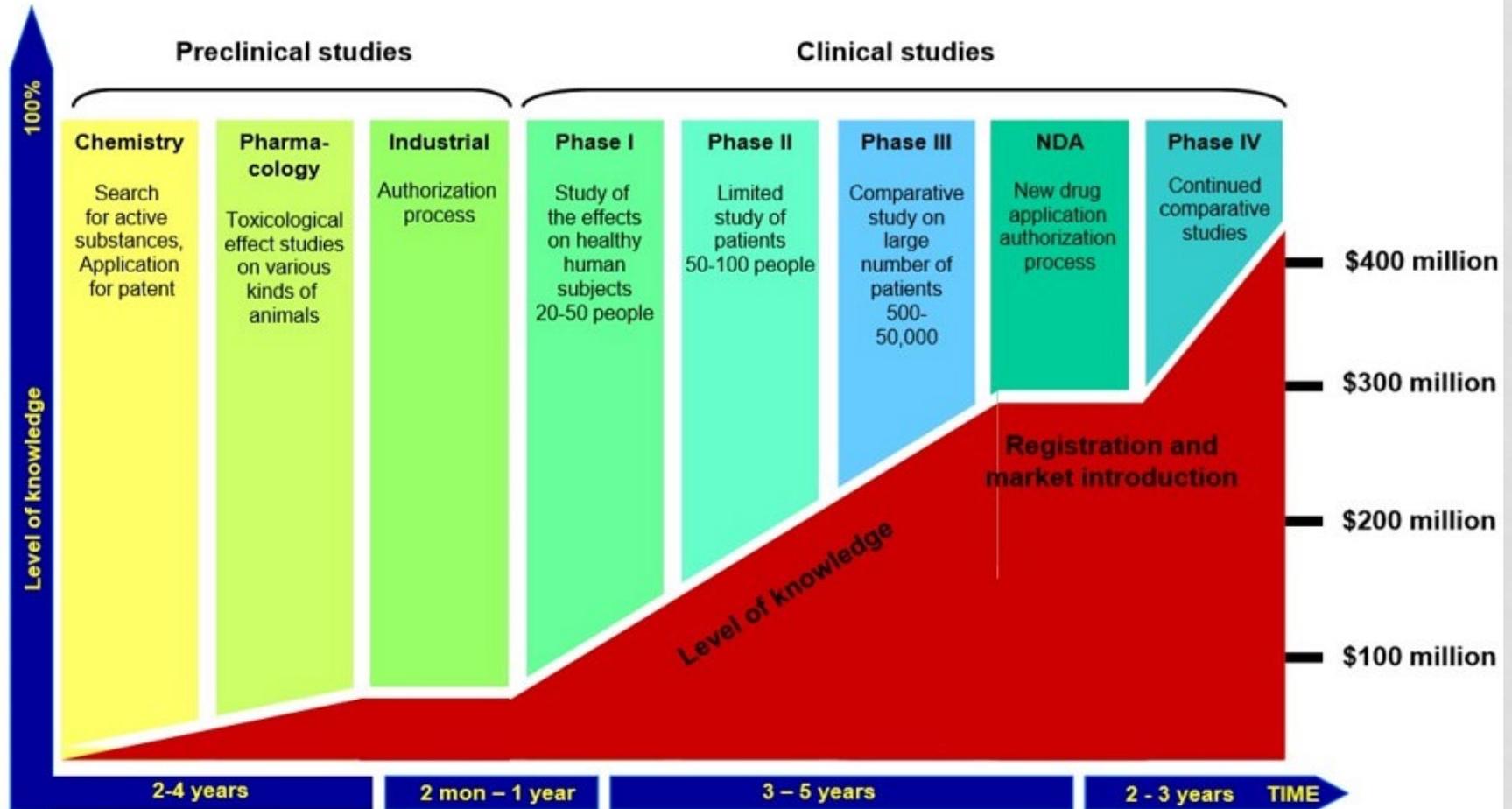


Key: IND=Investigational New Drug Application, NDA=New Drug Application, BLA=Biologics License Application

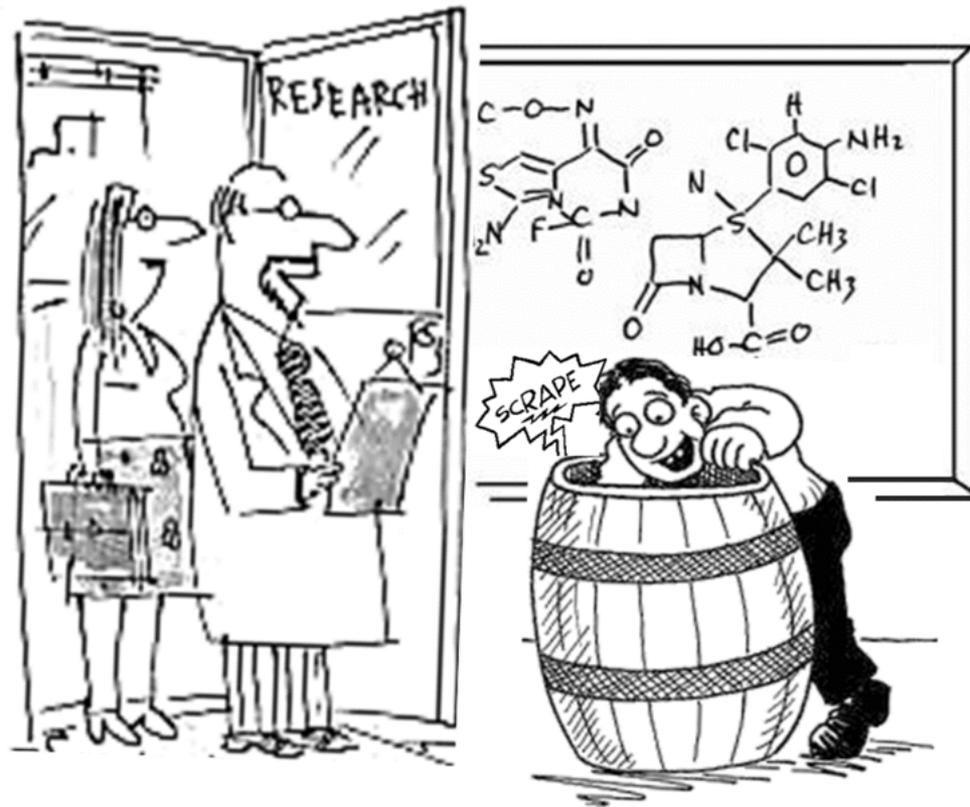
Drug Discovery Costs



Drug Discovery Costs



Why is it so expensive? All the “easy” drugs have been found already.

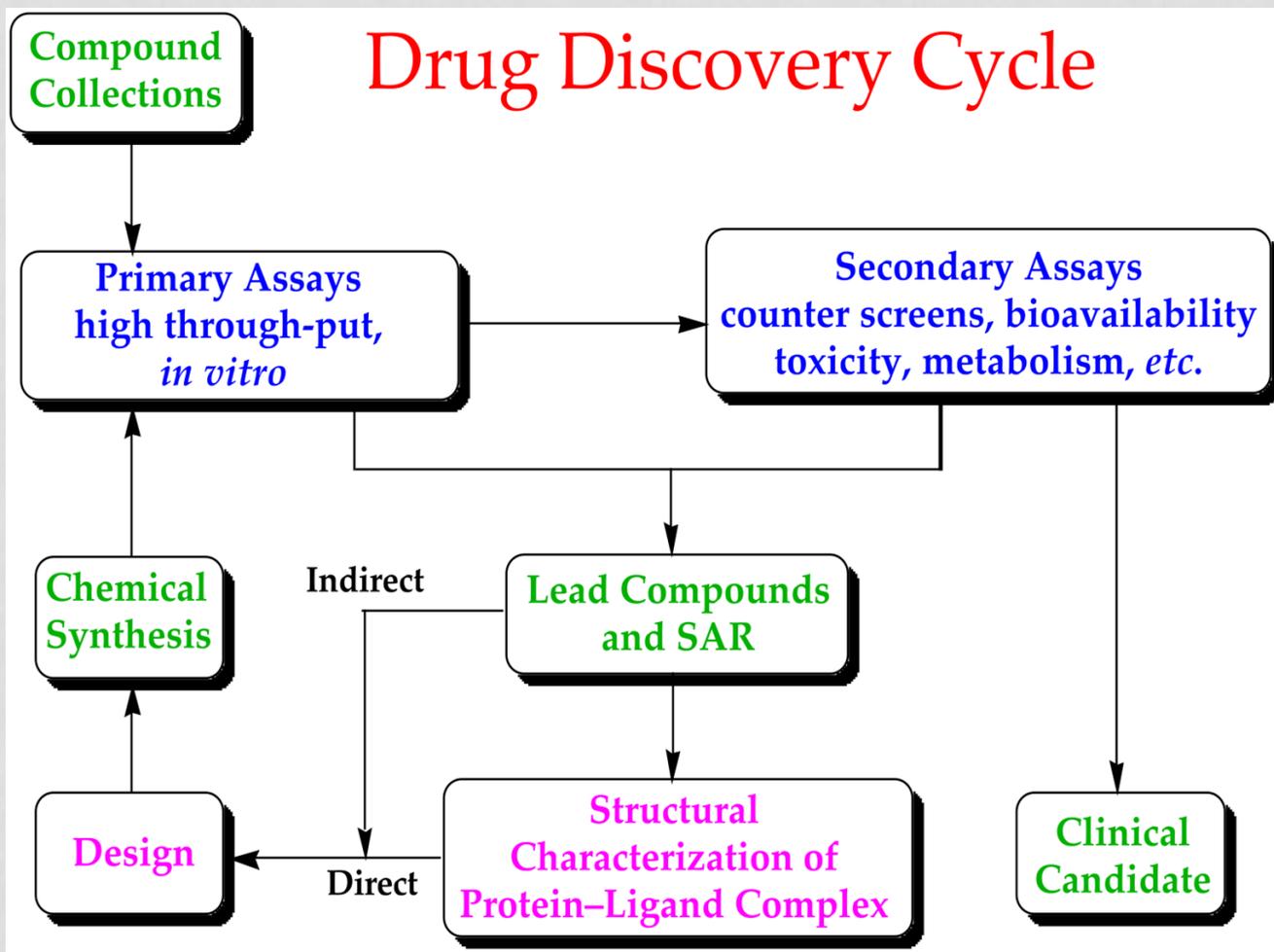


“And this is where we find all our new drug targets.”

What is a Drug Exactly?

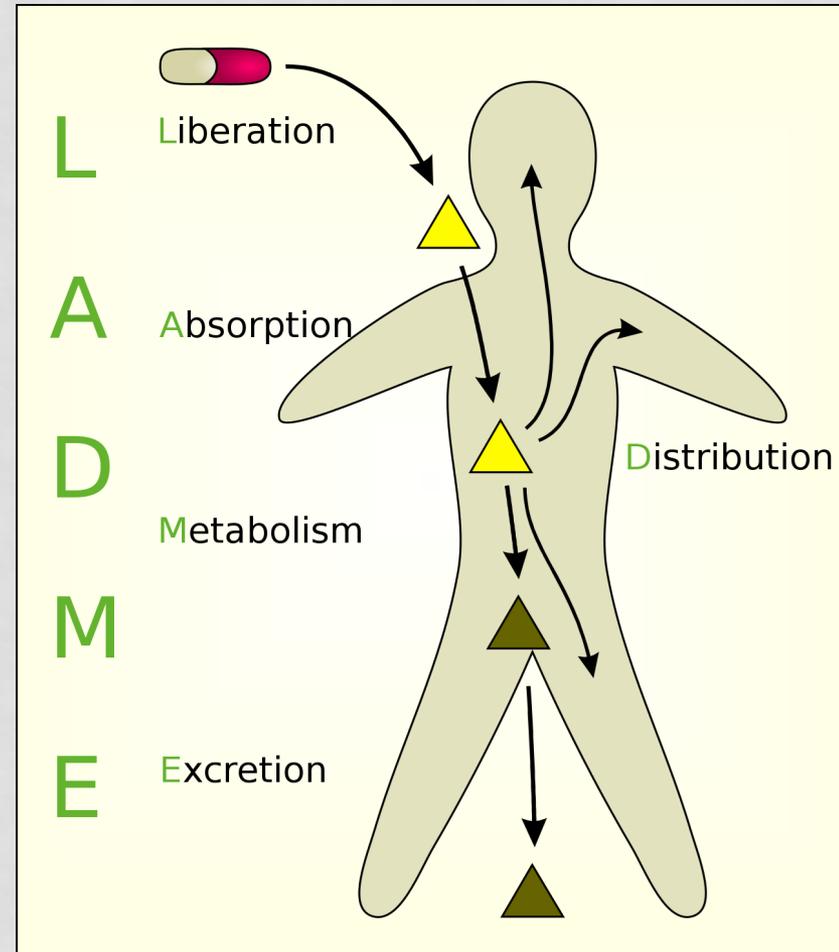
- What do you think a drug does?
- If you were going to take a medicine to make your headache go away, what would you want this medicine to do?
- Drug: a medicine or other substance which has a physiological effect when ingested or otherwise introduced into the body. (Oxford Languages)
- What if the drug does other things too?
- Are these effects good or bad?
- What are these effects called?

How Drug Discovery Works Without the Computers

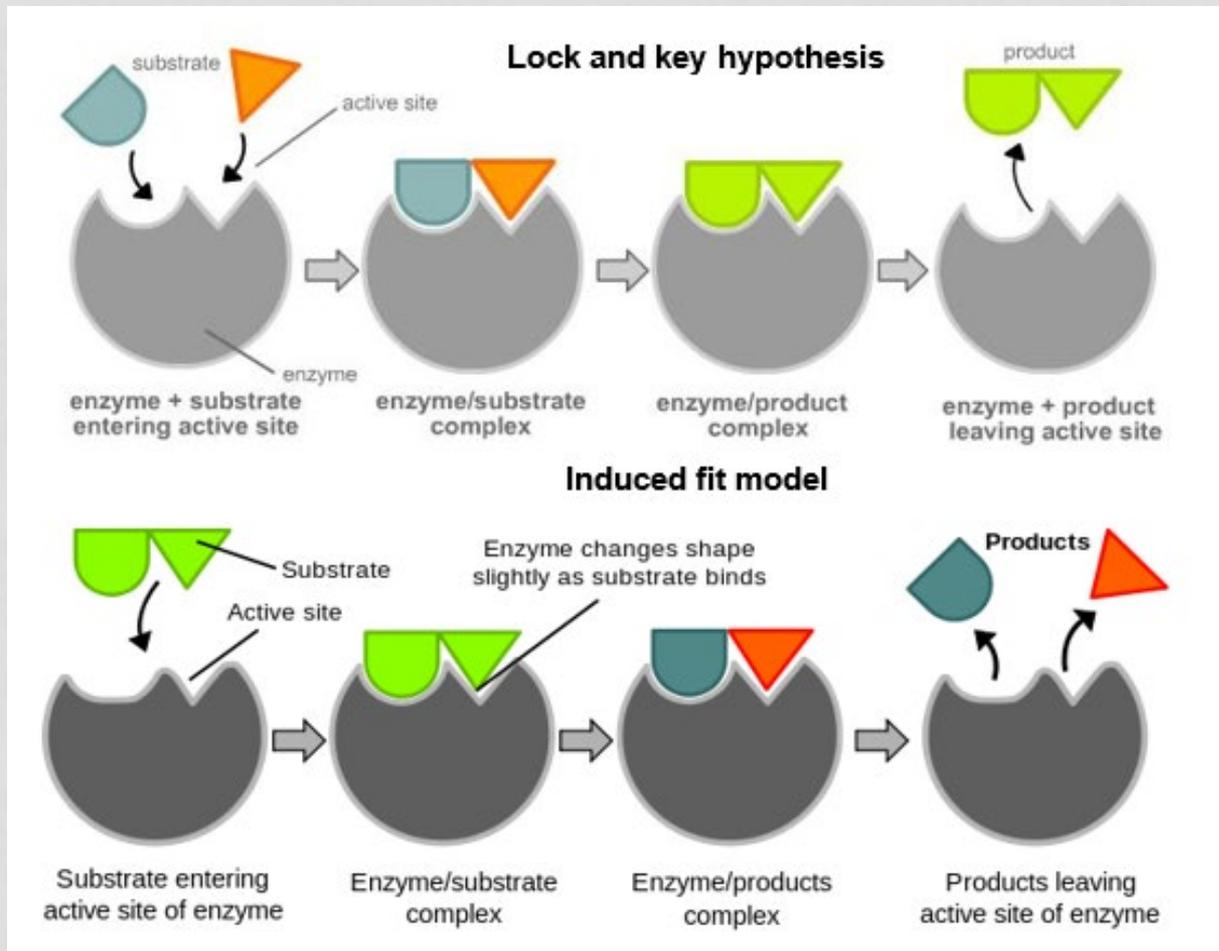


Kinds of Drug Design

- Drug design is all about design of a molecule i.e. the ligand or drug that will bind tightly to its *intended* target, usually the protein, that also follows (L)ADME(T)
 - is bioavailable,
 - has appropriate metabolic half-life,
 - is so selective it does not bind to other proteins i.e. no side effects,
 - is not **toxic**



Lock and Key Models

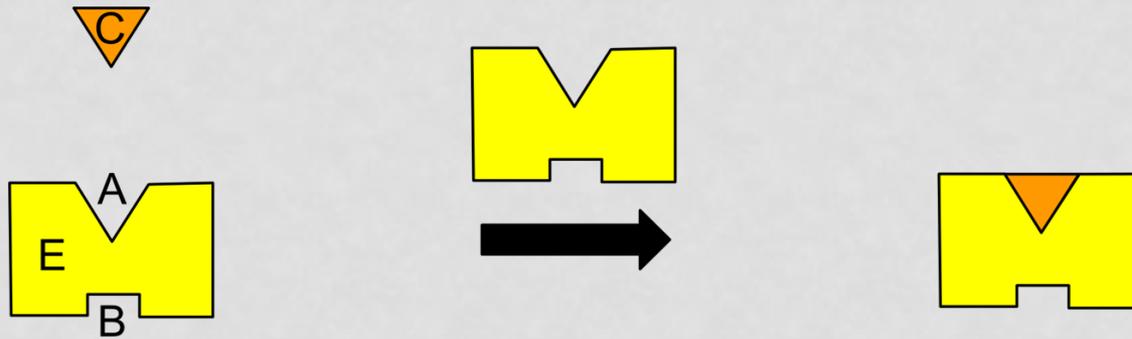


also conformational selection

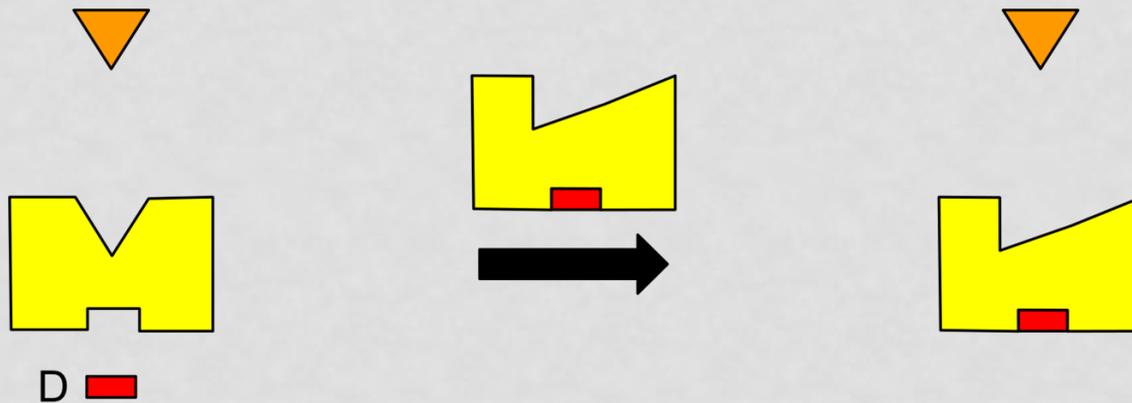
<https://alevelbiology.co.uk/notes/biological-catalysts-enzymes/>

Allosteric Interactions

1



2



Random Approach to Drug Discovery

Drug Discovery

One way to “discover” drugs



‘That’s Dr Arnold Moore. He’s conducting an experiment to test the theory that most great scientific discoveries were hit on by accident.’

*Drawing by Hoff; © 1957
The New Yorker Magazine, Inc.*

Figure 1.1

Other Kinds of Drug Discovery

- Structure-based drug design: drug design that relies on the knowledge of the three-dimensional structure of the biomolecular target is known. (more expensive, more accurate)
- Ligand-based drug design: drug design that relies on knowledge of other molecules that bind to the biological target of interest. (less expensive, less accurate)

Where Do the Computers Come in?

- Hit identification using virtual screening (structure- or ligand-based design).
- Hit to lead (H2L, lead generation) is a stage in early drug discovery where small molecule hits from a high throughput (virtual) screen (HT(V)S) are evaluated and undergo limited optimization to identify promising lead compounds.
- Hit-to-lead optimization of affinity and selectivity (structure-based design, QSAR, etc.)
- Lead optimization of other pharmaceutical properties while maintaining affinity

https://en.wikipedia.org/wiki/Drug_design

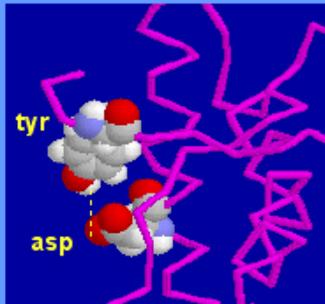
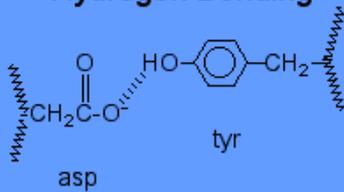
https://en.wikipedia.org/wiki/Hit_to_lead

Refresher: Tertiary Structure

- The tertiary structure is the final specific geometric shape that a protein assumes.
- There are four types of bonding interactions between "side chains" including
 - hydrogen bonding
 - salt bridges
 - disulfide bonds
 - non-polar hydrophobic interactions
- Often what drives more favorable binding is the molecule ejecting more unstable water molecules from the protein pocket while maintaining these properties.

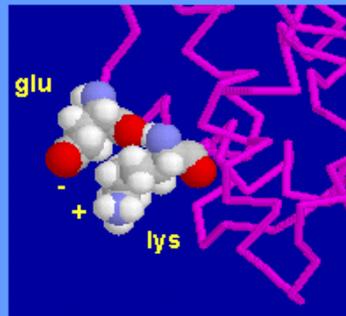
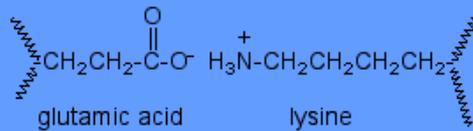
Polar Interactions

Tertiary Structure - Hydrogen Bonding



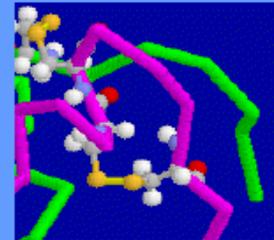
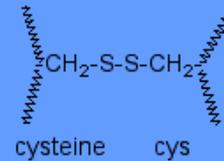
C. Ophardt, c. 2003

Tertiary Structure - Salt Bridges

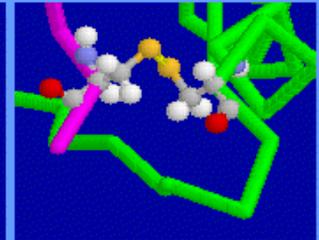


C. Ophardt, c. 2003

Tertiary Structure - Disulfide Bonds



Loop in single chain



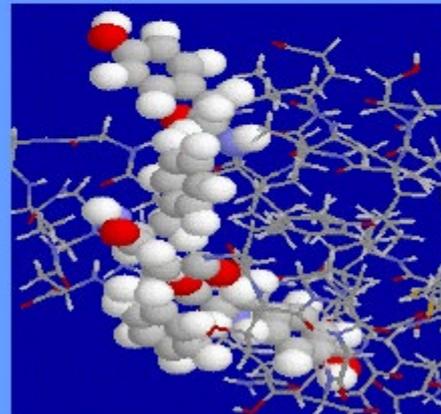
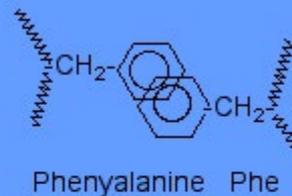
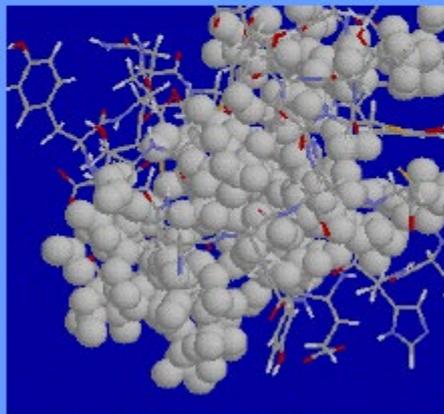
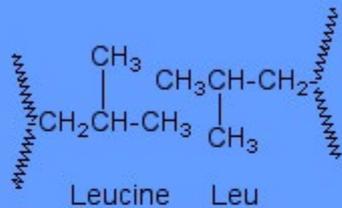
Join two chains

C. Ophardt, c. 2003

- Tyr is **donating** a hydrogen bond to Asp which is **accepting**

Nonpolar Interactions

Tertiary Structure - Hydrophobic Interactions



C. Ophardt, c. 2003

How Hydrophobic is the Drug?

Octanol-Water Partition Coefficient

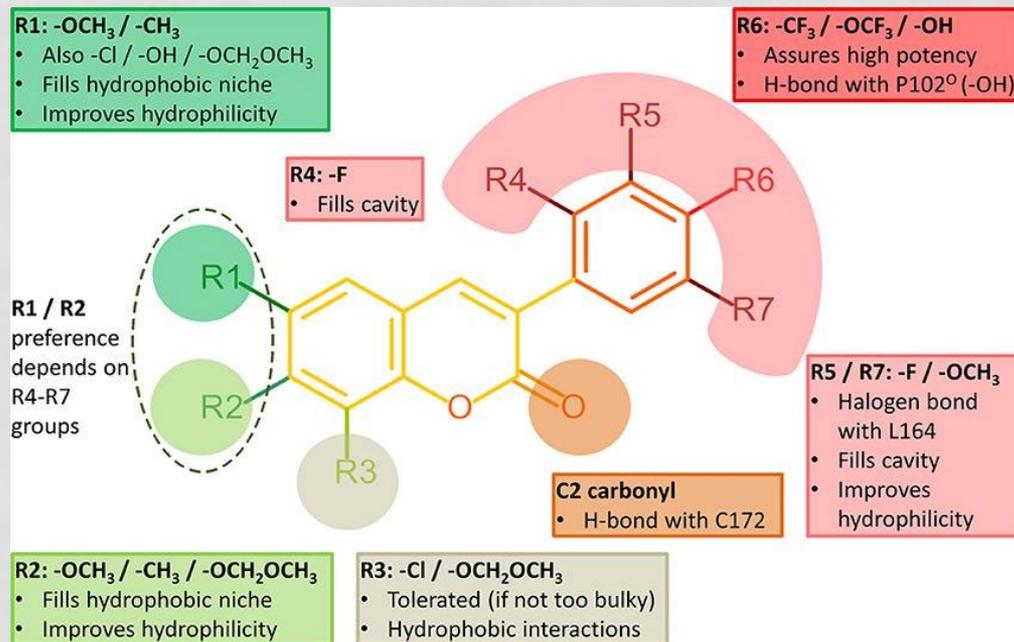
- The partition coefficient measures how hydrophilic ("water-loving") or hydrophobic ("water-fearing") a chemical substance is.
- Partition coefficients are useful in estimating the distribution of drugs within the body.
- Hydrophobic drugs with high octanol-water partition coefficients are mainly distributed to hydrophobic areas such as lipid bilayers of cells.
- Conversely, hydrophilic drugs (low octanol/water partition coefficients) are found primarily in aqueous regions such as blood serum.

Lipinski's Rule of Five

1. No more than **5** hydrogen bond donors (the total number of nitrogen–hydrogen and oxygen–hydrogen bonds).
2. No more than **10** hydrogen bond acceptors (all nitrogen or oxygen atoms).
3. A molecular mass less than **500** Daltons (molecular weight units)
4. An octanol-water partition coefficient ($\log P$) that does not exceed **5** (becoming too hydrophobic).

Structure-Function Relationships

- (Quantitative) structure–activity relationship = (Q)SAR
- Relationship between the chemical structure of a molecule and its biological activity.
- Modification of the effect or the potency of a bioactive compound (typically a drug) by changing its chemical structure using chemical synthesis.

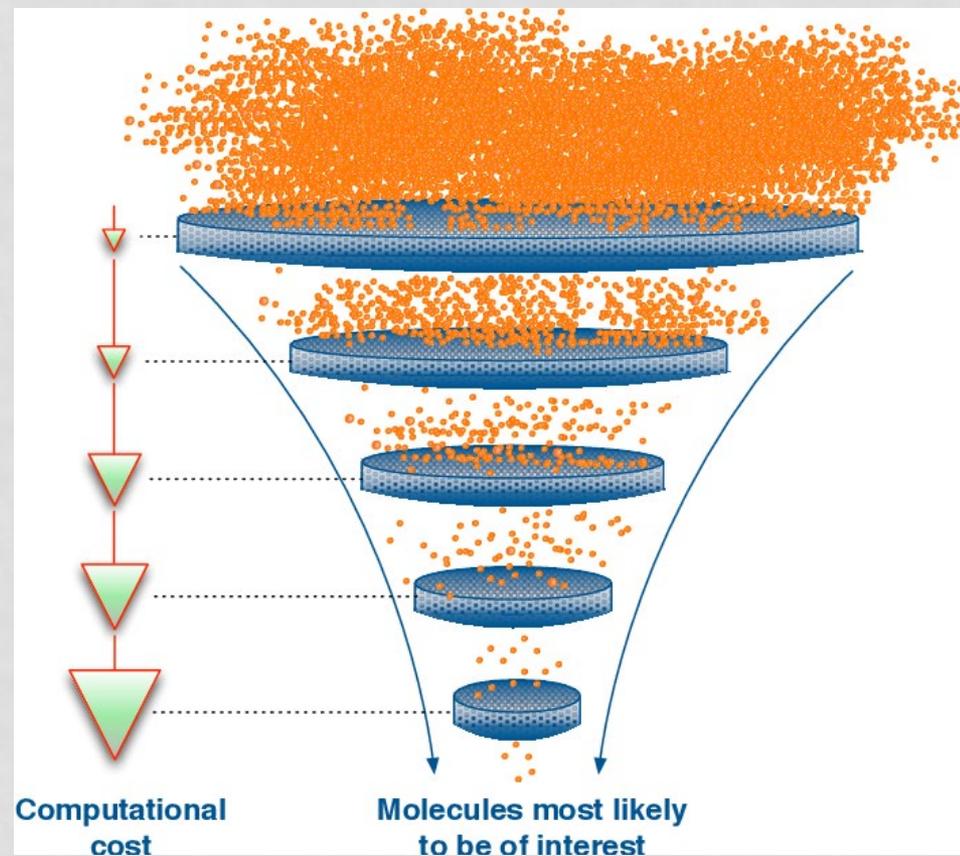


<https://www.frontiersin.org/articles/10.3389/fchem.2018.00041/full>

[https://en.wikipedia.org/wiki/Structure–activity_relationship](https://en.wikipedia.org/wiki/Structure%E2%80%93activity_relationship)

High-Throughput Virtual Screening

- Using progressively more expensive methods to screen through a large library of possible drug molecules to decide which are worth testing with the most expensive theoretical methods/experiments



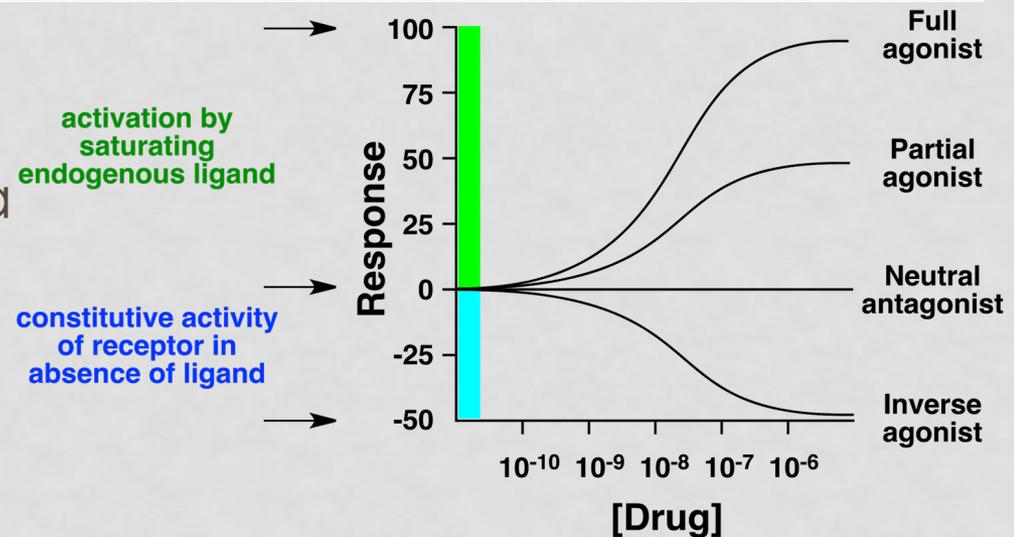
<https://www.annualreviews.org/doi/abs/10.1146/annurev-matsci-070214-020823>

<https://www.semanticscholar.org/paper/What-Is-High-Throughput-Virtual-Screening-A-from-Pyzer-Knapp-Suh/e142490c28bb6b954632e85ed6001eb312d001e7>

Suh/e142490c28bb6b954632e85ed6001eb312d001e7

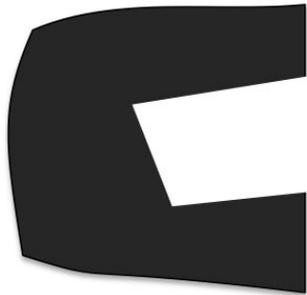
Agonist vs. Antagonist vs. Inverse Agonist

- Agonist: binds to a receptor and activates the receptor to produce a biological response.
- Antagonist (blockers): blocks or dampens a biological response by binding to and blocking a receptor.
- Inverse agonists: binds to the same receptor as an agonist but induces a pharmacological response opposite to that of the agonist.



Drug Docking

Target



+

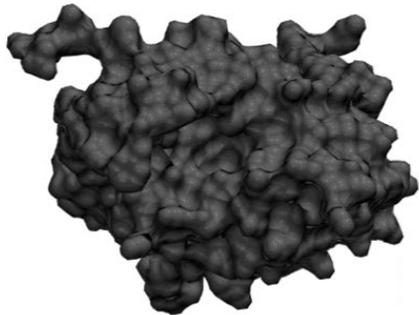
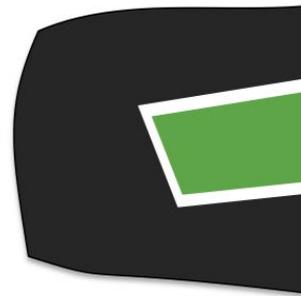
Ligand



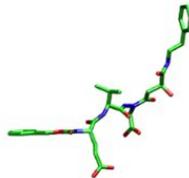
docking



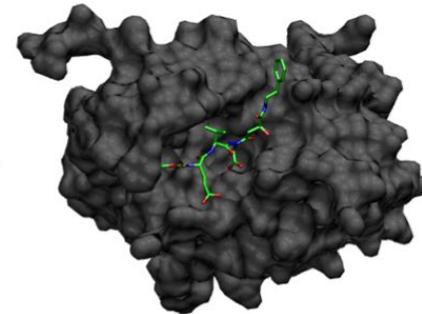
Complex



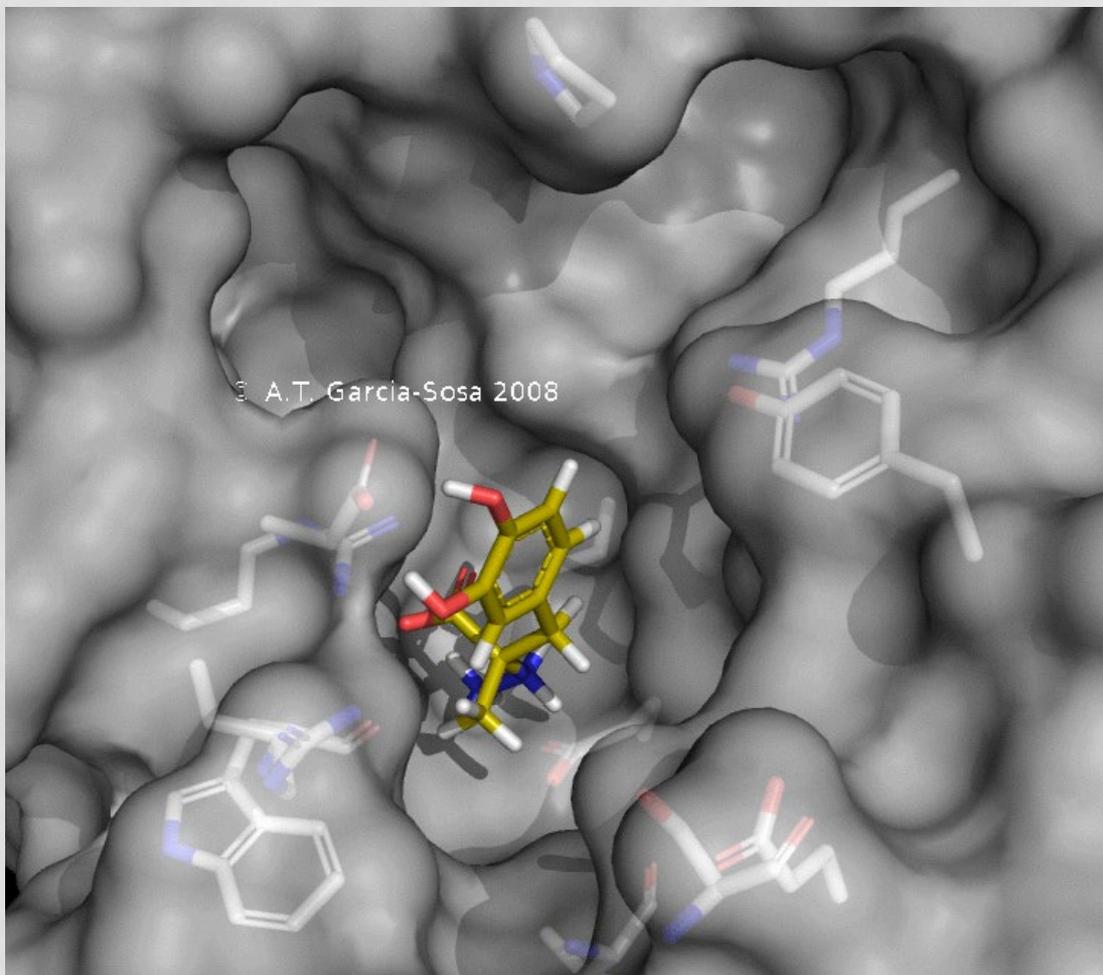
+



docking



Docking of compounds exploring the binding sites of H5N1 avian influenza neuraminidase.



Drug Docking: Typical Scoring Function

$$\Delta G_{\text{bind}} = -RT \ln K_d$$

$$K_d = \frac{[\text{Ligand}][\text{Receptor}]}{[\text{Complex}]}$$

$$\Delta G_{\text{bind}} = \Delta G_{\text{desolvation}} + \Delta G_{\text{motion}} + \Delta G_{\text{configuration}} + \Delta G_{\text{interaction}}$$

where:

- desolvation – **enthalpic** penalty for removing the ligand from solvent
- motion – **entropic** penalty for reducing the degrees of freedom when a ligand binds to its receptor
- configuration – conformational strain energy required to put the ligand in its "active" conformation
- interaction – enthalpic gain for "resolvating" the ligand with its receptor

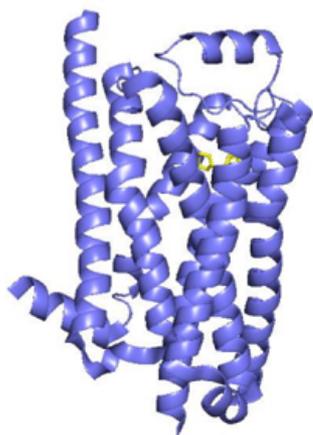
Docking and Decoys

- Test/train the docking algorithm on some target proteins with known strong binding molecules
- “An imperfect scoring function can mislead by predicting incorrect ligand geometries or by selecting nonbinding molecules over true ligands. These false-positive hits may be considered ‘decoys’.”
- Docking can be static or induced fit.

Video:

https://commons.wikimedia.org/w/index.php?title=File%3ADocking_GPCR_example.webm

What if Structure Has to Relax?



Crystal Structures

Protein Structure Databank (PDB)



Molecular
Dynamics
Simulations



New Binding Sites



Free Energy

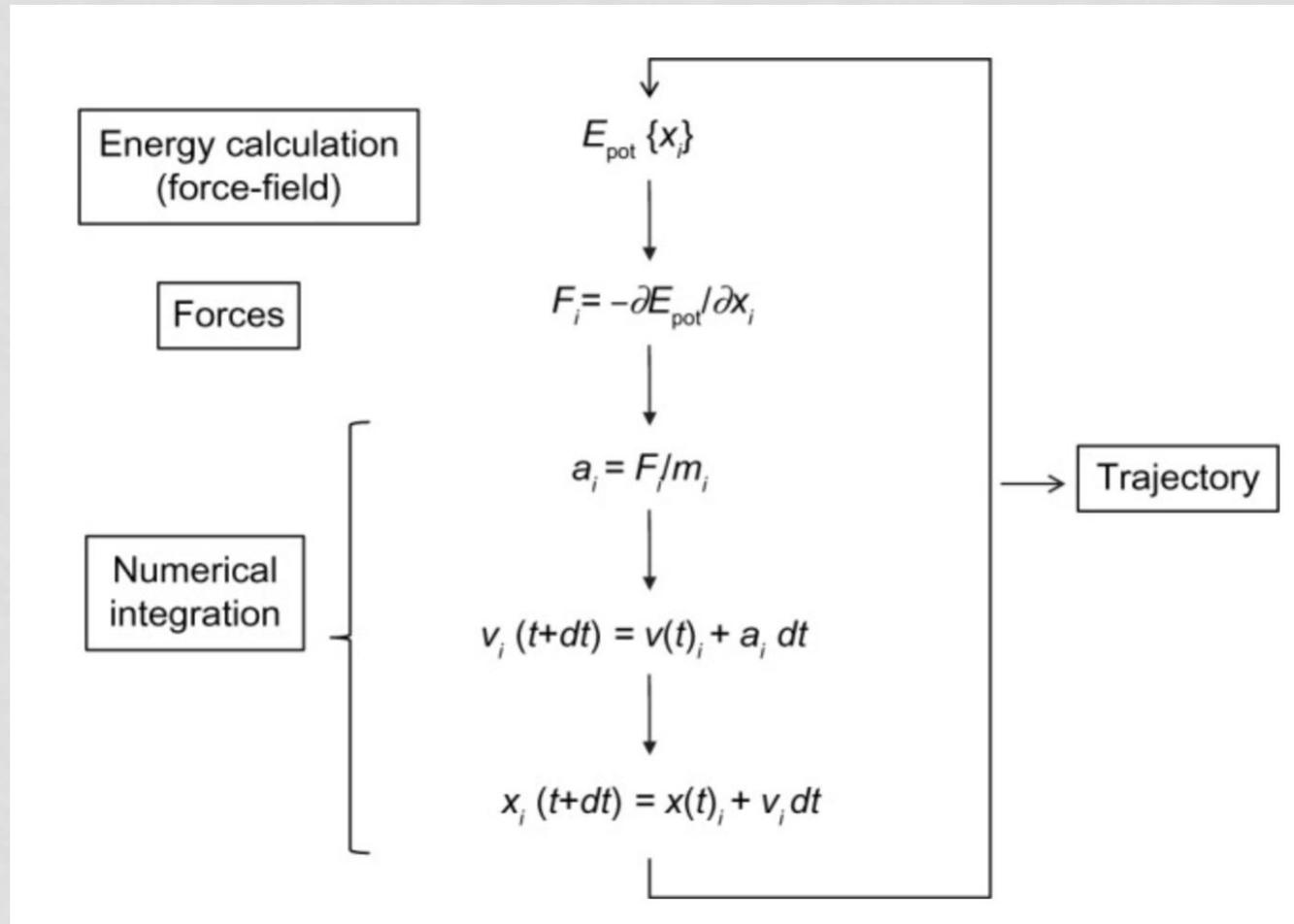


Dynamics



Drug Design

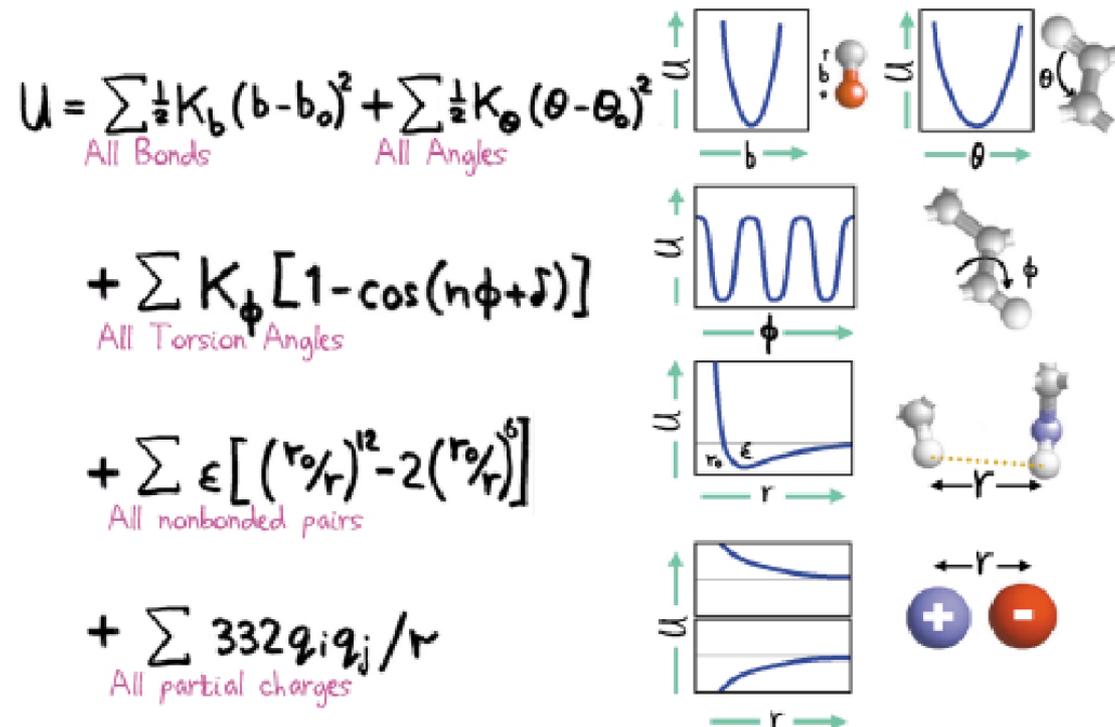
Molecular Dynamics (MD): Classical Physics with Special Potential Energy Function



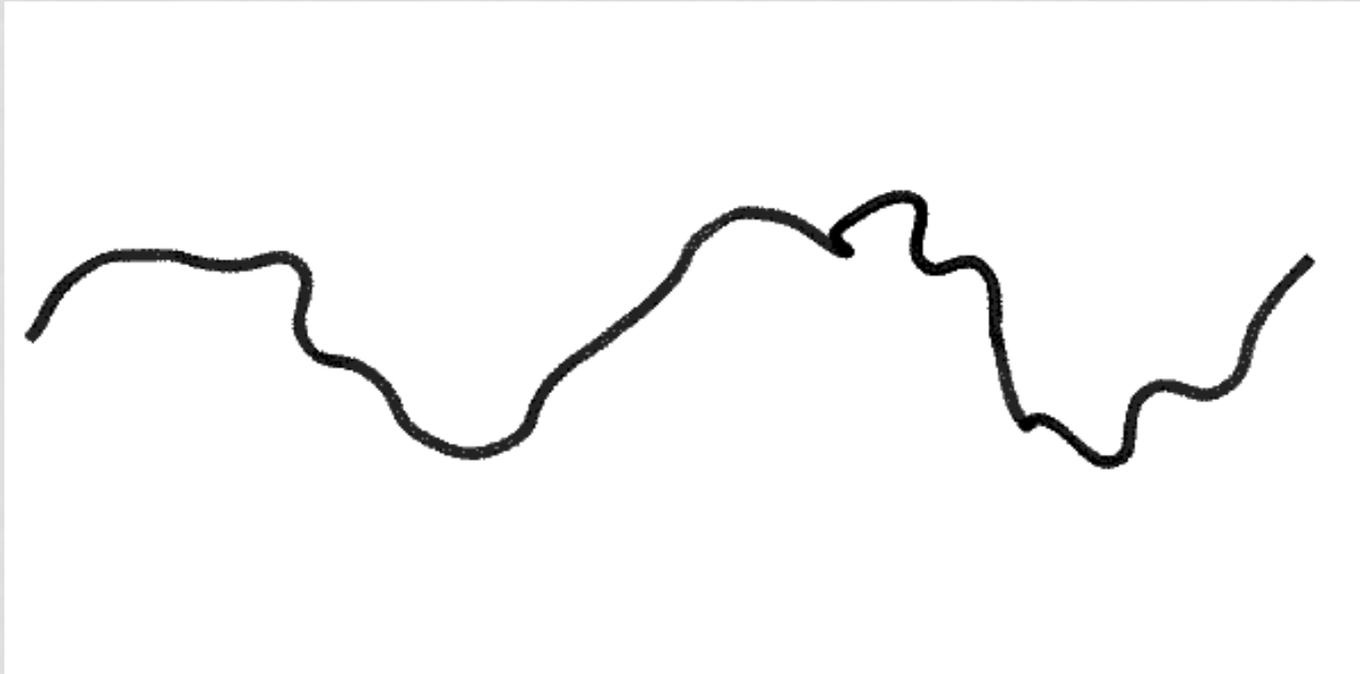
Potential Energy Function

Levitt

Nat. Struc. Bio. 2001, 8, 392

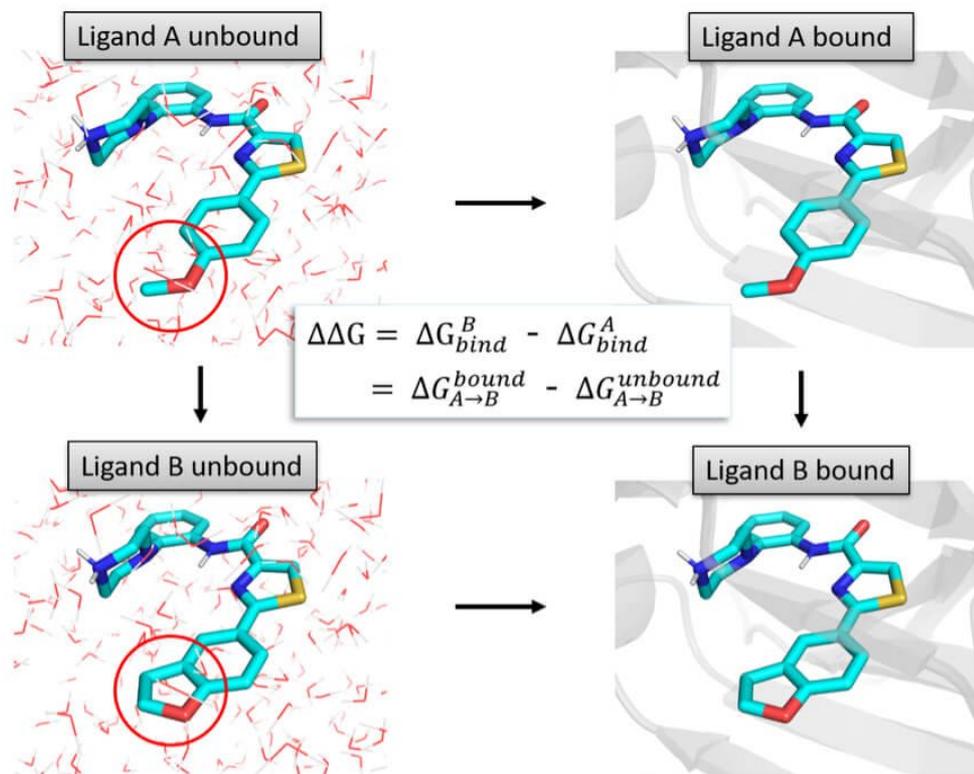


Protein Folding Example



<https://medium.com/proteinquire/welcome-into-the-fold-bbd3f3b19fdd>

Free Energy Perturbation (FEP)



Conclusions

- Drug discovery is very expensive and laborious.
- Drug molecules have to not only bind very well to the protein (potency; focus of the CADD), but also satisfy many other (pharmacophore) characteristics.
- Computer-aided drug design (CADD) offers a cost-saving approach (in comparison to experiments) to identify which molecules are worth studying further (screening).
- The expense of calculations is often directly related to their accuracy.
- Hierarchy of methods: ligand-based screening < rigid docking (classic lock + key) < induced fit docking (modern lock + key) < molecular dynamics approaches
- These methods use aspects of tertiary structure like hydrogen bonds, ionic bonds, hydrophobic interactions, and disulfide bridges.

Questions?

Interested in getting involved in research?

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In terms of getting started with computational chemistry research, I would recommend:

1. Taking rigorous courses in high school
2. Getting involved in the Regeneron (formerly Intel?) science fair high school program if possible
3. Getting involved in research in college.
4. However, there are some pre-college programs in computational chemistry/biology like:
 - a. <http://cbd.cmu.edu/education/pre-college/index.html>
 - b. <http://www.discobio.pitt.edu/>
 - c. <https://cns.utexas.edu/tides/k-12/high-school-summer-research-academy>